

Earth Science 11

Unit 2 – The Geology of Earth

Day 1 – Earth's History

Name: Schaub

Date: _____

Block: _____

Geologist/Geophysicist/Earth Scientist: Primary goal is to interpret Earth's history.

Rocks record geological events and evidence of ancient life forms.

Uniformitarianism: Geological processes of today are similar to those of the past.

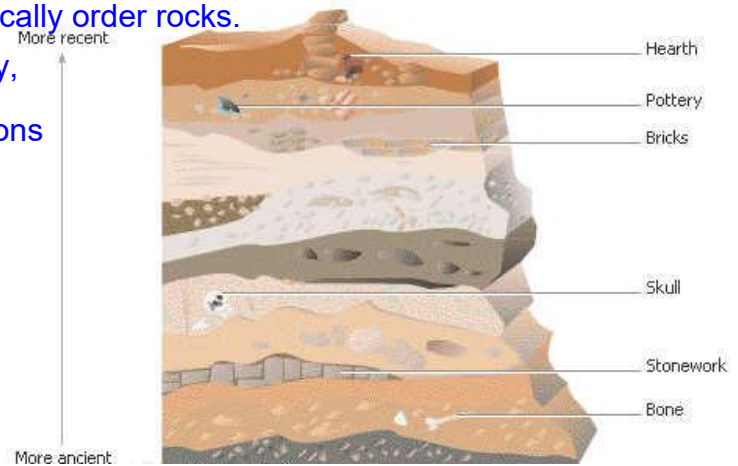
"The present is the key to the past."

Relative Dating: Method geologists used to chronologically order rocks.

Law of superposition, principle of original horizontality,
principle of cross-cutting, unconformities, and inclusions
all help determine the relative ages of rock layers

Law of Superposition: _____

Sequence of sedimentary rock layers with younger
rocks on top of older rock layers

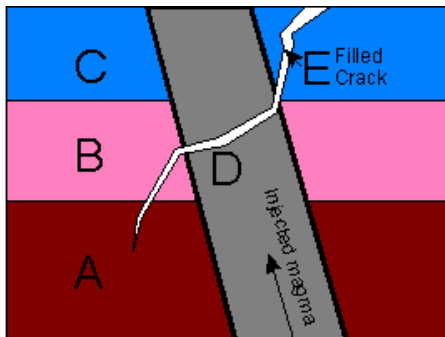
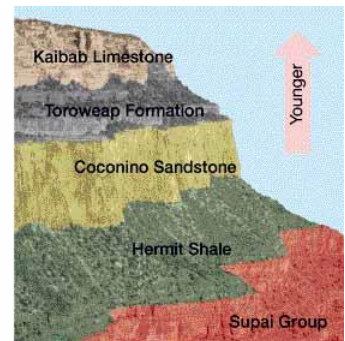


Principal of Horizontality: _____

Sediments are deposited in relatively horizontal layers
Erosion and gravity tends to flatten everything down
over time.

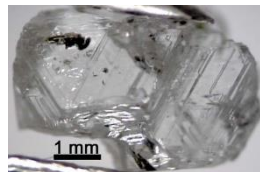
Principal of Cross-Cutting: Magma intrusions are younger
than the rocks they cut through.

Occurs near present and past geological active zones



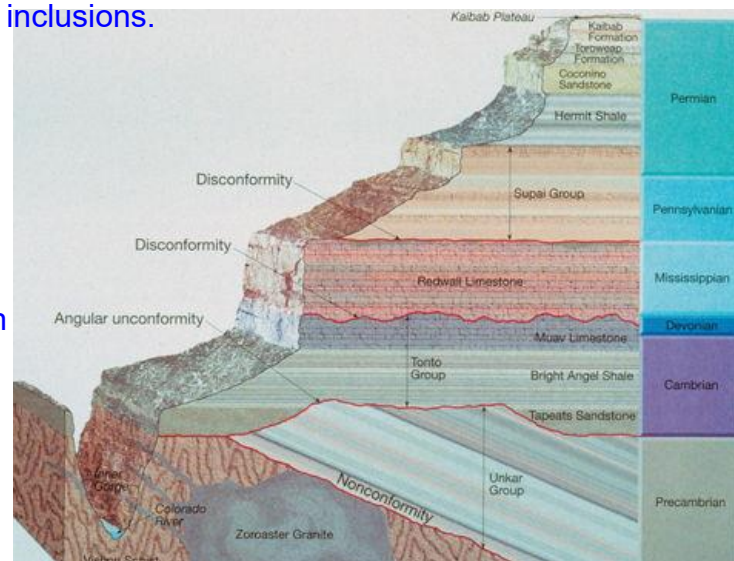
Inclusions: Pieces of rock inside another are older than the rock they are in.

Example is diamond inclusions.



Unconformities: Break in the rock record due to erosion
or lack of deposition

Geologists determine, which time intervals are not
present in the rock layers



What, when, where, how?

Fossils: Remains, imprints, or traces of once-living organisms preserved in rock

Fossils are formed by the conditions in which they died and how it was buried.

Radioactivity: Unstable atomic nuclei spontaneously break apart, or decay, releasing energy

Three types of radioactive decay: Alpha emission
Beta emission
Electron capture

Half Life: Amount of time necessary for one half of the nuclei in a sample to decay to its stable isotope

Isotope: Form of an element (varies in the number of neutrons).

Carbon-14 Dating: When an organism dies, the amount of C-14 gradually decreases and decays.

The ratio between Carbon-14 and Carbon-12 informs scientists the amount of half-lives.

Half-life of C-14 is 5730 years. Used to date geological events up to 75,000 years

